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'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of a liquid placed in. a container, equilibrium is established when the Equid vaporizes at the same rate that vapor condenses. 7, NO, in reversible reactions at equilibrium,

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## CHAPTER 16

II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is  $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$ . 14. Aluminum reacts with HCl to produce aluminum chloride ( $\text{AlCl}_3$ ) and hydrogen ...

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