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molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a. molar mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg SO_3 1 g 1000 mg 1 mol $80.07 \text{ g} = 6.14 \cdot 10^{-4}$ mol SO_3 b. molar mass of $PbO_2 = 239.2 \text{ g}$ $7.44 \times 10^4 \text{ kg}$ PbO_2 1000 g 1 kg 1 mol $239.2 \text{ g} = 3.11 \cdot 10^5$ mol PbO_2 c. molar mass of $CHCl_3 = 119.37 \text{ g}$ 59.1 g $CHCl_3$ 1 mol $119.37 \text{ g} = 0.495$ mol $CHCl_3$ d. molar mass of $C_2H_3Cl_3 = 133.39 \text{ g}$ 3.27 ...

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MHz. (1 MHz 10^6 s⁻¹) What is the wavelength of the radiation emitted by the station?

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Answer to 12. Each period in the periodic table corresponds to .
a. an energy level c. an orbital b. an electron d. a suborbital

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