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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the Textbook)

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= 12.00 mmol - 5.40 mmol Unreacted ethanoic acid = 6.60 mmol Since 18.00 mL of NaOH(aq) was added to 20.00 mL of ethanoic acid solution, the total volume is now 38.00 mL. Solution: The concentration of unreacted ethanoic acid, [HC 2H 3O 2(aq)], can be determined: 23 2 23 2 HC H O (aq) 23 2 HC H O (aq) 23 2 [HC H O (aq)] (6.60 mmol) = 38.00 mL

Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547

2 Chemistry, First Canadian Edition 1.17 An organic compound consists of molecules made from carbon, hydrogen and oxygen atoms. There can be other atoms - for example, nitrogen, sulfur. 1.19 1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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8 MHR Chemistry 12 Solutions Manual 978-0-07-106042-4 8. Identify any errors in the structure by drawing them. Rename the structure correctly. 2-ethyl-2,4,4-trimethylpentane What is Required? You are to identify the errors in the name by drawing the structure and renaming it correctly. What is Given? You are given an incorrectly named alkane.

Unit 1 Organic Chemistry - Mr. Arthur's Science Page

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