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Stoichiometry. study of quantitative relationships between the amounts of reactants used and the amounts of products formed in a chemical reaction. Mass of Reactants = Mass of Products. ...
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Chapter 11 and 12 Study Guide Molarity Molality moles of solute
L of solution moles of solute kg of solvent Percent mass grams of

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solute x 100 grams of solution Mole Fraction mol solute total moles Density of Solution grams solution mL solution X Solution Formation Electrolytic ions present in solution Hydrated ions. interactions keep the ions from recrystallizing Not everything that dissolves in water is an electrolytic solution. o Sugar dissolves.

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Xe, Po 2. How many core electrons are there in an Sb ($Z = 51$) atom?

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