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3) 3 is an isomer of X. For the 1H NMR spectrum of this isomer, deduce the total number of peaks (excluding the TMS peak at 0 ppm) and the ratio of peak areas. For each peak, deduce whether it is a singlet, doublet, triplet, quartet or shows a more complex splitting pattern. [3] Number of peaks (excluding the TMS peak at 0 ppm):

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Many recent developments in chemistry have involved making use of devices that operate on a nanoscale. (a) (i) State the scale at which nanotechnology takes place and outline the importance of

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M11/4/CHEMI/HP3/ENG/TZ1/XX A4. Phenolphthalein is colourless at pH 7 but its structure changes at pH 11 and it becomes pink.

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(a) vapour pressure ethoxyethane (81 10 Pa× 3) > vapour pressure benzene (16 10 Pa× 3) > vapour pressure water (4 10 Pa× 3); If three correct vapour pressure values related to each substance are stated alone award M1. Allow range of 80 85 10 Pa, 14 18×10 Pa and 3 7×10 – – – .× 33 3 Pa

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(v) 25.0 cm3 of 1.00 10 moldm× − −2 3 hydrochloric acid solution is added to 50.0 cm3 of 1.00 10× −2 mol dm –3 aqueous ammonia solution. Calculate the concentrations

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hl CHEMISTRY NOTES. TOPIC 1: QUANTITATIVE CHEMISTRY. TOPIC 2: ATOMIC STRUCTURE ... 9: REDOX PROCESSES. TOPIC 10: ORGANIC CHEMISTRY. TOPIC 11: MEASUREMENT AND DATA PROCESSING. OPTION B: BIOCHEMISTRY. THE HELPFUL CHEMISTRY RESOURCES I FOUND USEFUL. RadioChemistry: IB Online Teacher; ... past papers. specimen papers. Powered by Create your own ...

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Question 3 (5 marks) in the 2018 May Time Zone 1 (TZ1) Maths HL Paper 1 Exam is a Probability distribution question where a table is given with two unknown probabilities.

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